

## REVIEW

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## Mini Review: Clincinal Strategies for Carbon monoxide Poisoning

Kısa Derleme: Karbonmonoksit Zehirlenmesinde Klinik Stratejiler

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## ABSTRACT

Carbon monoxide is a type of tasteless, colorless, odorless gas that occurs as a result of incomplete combustion of organic/inorganic hydrocarbons. He is defined as the "silent killer" based on these three physical characteristics. Poisoning is seen especially in winter months due to chimney systems used in rural areas (poisoning caused by suicide and exhaust may also be seen). In a study conducted in our country, it was shown that most of the exposure to carbon monoxide gas between 1993 and 2006 was caused by stoves and water heaters without any suicidal intent. It has been stated that the affinity of carbon monoxide to the tetramer receptor region in hemoglobin is approximately 200-250 times higher than that of the oxygen molecule, resulting in hypoxia and relative anemia. Additionally, by binding to cardiac myoglobin, it causes myocardial depression and hypotension. It plays a role in the activation of platelets, causing myeloperoxidase release, high oxidative stress, inflammation, and deterioration of cellular respiration by binding and inactivating reduced cytochrome a3. Intoxication resulting from these mechanisms must be treated as soon as possible. Otherwise, life-threatening situations may occur. The first symptoms of carbon monoxide intoxication are non-specific. Physical examination findings have a limited place in diagnosis. The relationship between disease history and carboxyhemoglobin is the most reliable diagnostic tool. There is no chemical or physiological antidote for poisoning. In first aid, advanced life support steps must be applied depending on whether the patient is conscious or unconscious. Hyperbaric oxygen therapy dramatically increases patient survival and is now recommended as the gold standard treatment by separating carbon monoxide bound to hemoglobin.

**Keywords:** Carbon Monoxide, Hyperbaric Oxygen, Intoxication.

## ÖZET

Karbonmonoksit organik/inorganik hidrokarbonların tam olarak yanmaması sonucu ortaya çıkan, tatsız, renksiz, kokusuz bir gaz türüdür. Kendine ait bu üç fiziksel özelliğe binaen "sessiz katil" olarak tanımlanmaktadır. Zehirlenme özellikle kış aylarında kırsal kesimde kullanılan baca sistemleri nedeniyle görülmektedir (Öz kıyım ve egzoz kaynaklı zehirlenmeler de görülebilmektedir). Ülkemizde yapılan bir çalışmada, 1993-2006 yılları arasında karbonmonoksit gazına maruziyetlerin çoğunun intihar amacı güdülmeyen soba ve şofben kaynaklı olduğu gösterilmiştir. Karbonmonoksidin hemoglobindeki tetramer reseptör bölgesine olan afinitesinin oksijen molekülünden yaklaşık 200-250 kat daha fazla olduğu ve buna bağlı hipoksi, göreceli anemi geliştiği belirtilmiştir. Ayrıca kardiyak myoglobine bağlanarak myokardiyal depresyon, hipotansiyon, trombositlerin aktivasyonunda rol oynayarak myeloperoksidaz salımına, yüksek oksidatif strese, inflamasyona ve redükte sitokrom a3'ü bağlayıp etkisiz hale getirerek hücresel solunumun bozulmasına yol açmaktadır. Bu mekanizmaların sonucu ortaya çıkan intoksikasyonun tedavisinin en kısa sürede yapılması gereklidir aksi takdirde hastanın hayatını tehdit edebilecek durumlar ortaya çıkabilir. Karbonmonoksit intoksikasyonunun ilk belirtileri non-spesifiktir. Fizik muayene bulgularının tanıda yeri sınırlıdır. Hastalık öyküsü ve karboksi-hemoglobin arasındaki ilişki en güvenilir tanı aracı olmaktadır. Zehirlenmenin kimyasal veya fizyolojik bir antidotu bulunmamaktadır. İlk yardımda hastanın bilincinin açık ya da kapalı olma durumlarına göre ileri yaşam desteği basamaklarının uygulanması gereklidir. Hastada sağ kalımı yüksek oranda arttıran hiperbarik oksijen tedavisi hemoglobine bağlı karbonmonoksidi ayırarak günümüzde altın standart tedavi olarak önerilmektedir.

**Anahtar Kelimeler:** Hiperbarik Oksijen, İntoksikasyon, Karbonmonoksit.

## INTRODUCTION

Carbon monoxide is a simple chemical in structure but complex in application. Structurally, it is simple, like nitrogen gas (N<sub>2</sub>). It consists of two atoms, has the same number of electrons, and has a triple bond between its atoms. Moreover, although the boiling points of the two gases are close to each other, liquid nitrogen boils at -196 °C, and liquid carbon monoxide boils at -192 °C. The asymmetry that makes carbon monoxide a polar molecule reveals its complex structure (1,2).

Carbon monoxide occurs due to incomplete combustion of organic/inorganic hydrocarbons, and incompletely burned natural gas is also a source of carbon monoxide. Although it is a tasteless, colorless,

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odorless gas, poisoning is not noticed. He is defined as the "silent killer" based on these three physical characteristics (3). Since it is a very toxic gas, it is among the most common causes of fatal poisoning (4).

Among the 350 million population of the USA, which is among the developed countries, an average of 45 thousand people is affected by carbon monoxide poisoning every year, and 4 thousand of them die due to carbon monoxide. Mortality/morbidity data obtained from studies conducted in the USA clearly show how serious this type of poisoning has reached (5).

Although the studies conducted in our country are limited, in the study conducted in 2010, *10,154 carbon monoxide poisoning cases were identified from the records in Turkey. 39 of these cases resulted in death* (6).

## OBJECTIVE

This study aims to compile pharmacotherapeutic methods that can reduce the high mortality/morbidity rates of carbon monoxide intoxication and provide treatment based on the information in the literature.

## METHOD

This study has been researched, and articles relevant to studies from current medical and health databases have been examined.

## RESULTS

In our country, the source of poisoning appears to be from unsuitable, unmaintained, old waste gas removal parts in homes. In addition, home accidents, which increase in winter months or windy weather due to the chimney systems used primarily in rural areas, constitute a large part of poisonings (7). A 14-year study conducted by Dokuz Eylül University Faculty of Medicine in our country between January 1993 and December 2006 revealed that most of the exposure to carbon monoxide gas was caused by stoves and water heaters, without intending suicide (8).

### *Pathogenesis*

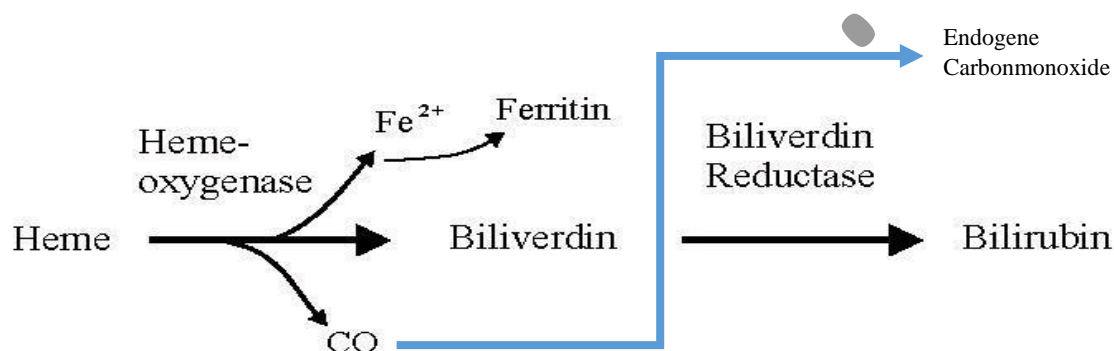
*Since carbon monoxide naturally occurs in gaseous form, inhalation is the main route of exposure in cases of poisoning. Carbon monoxide gas inhaled by the individual quickly and easily reaches the lower respiratory tract, which begins with the larynx. Carbon monoxide is absorbed in the alveoli, which are the extreme point of respiration and are known as the air sacs. Absorbed carbon monoxide alveolar It passes through the membrane and enters the intravascular space where it binds to hemoglobin. After carbon monoxide accumulates in erythrocytes, it creates a toxic effect characterized by hypoxia* (9).

*The amount of gas absorbed and, therefore, the toxicity status depends on the respiratory rate (air exchanges per minute), the duration of exposure, and the carbon monoxide and oxygen concentrations* (9).

*Carbon monoxide toxicity is its affinity for the body protein hemoglobin. Hemoglobin is a tetramer with four oxygen binding points. Carbon monoxide's affinity for the tetramer receptor site is approximately 200-250 times greater than the oxygen molecule's (10). This high-affinity carbon monoxide is thought to bind to hemoglobin and affect oxygen-carrying capacity by two mechanisms. The first mechanism is based on competitive inhibition of oxygen binding to hemoglobin, thus preventing oxygen from being transported and released into the tissues. This leads to a relative anemia caused by carbon monoxide, causing asphyxia or anemic hypoxia (11,12). Secondly, carbon monoxide causes structural changes with its inhibitory effect on cellular and proteins (such as in myoglobin and hemoglobin), making it difficult to deliver oxygen to the tissues. Reduced It disrupts cellular respiration by binding and inactivating*

cytochrome a3. It causes cells and tissue necrosis by reducing tissue energy production and preventing the necessary oxygen from reaching them (13).

Carbon monoxide has an affinity for cardiac myoglobin among cells, so myocardial depression and hypotension may occur due to tissue hypoxia. In addition to cardiac myoglobin, carbon monoxide also binds to linear muscle myoglobin, causing a decrease in partial oxygen pressure in muscle tissue, resulting in rhabdomyolysis, characterized by muscle pain (14,15). *Activated platelets, indirect myeloperoxidase release, high oxidative stress, and inflammation* (16). *The released myeloperoxidase and reactive oxygen species lead to the formation of myelin essential protein in cells. Myelin basic protein is also used as a diagnostic method* (17).



**Figure 1.** Formation Mechanism of Endogenous Carbon Monoxide (1).

## Clinic

Carbon monoxide poisoning is called the "disease with a thousand faces," and the symptoms of poisoning are similar to viral infections. Considering that both diseases peak in the winter months, making a differential diagnosis in cases seen in these seasons becomes challenging. For this reason, disease history becomes significant in the clinic. A possible history of chimney, stove, exhaust, or workplace exposure will suggest carbon monoxide poisoning. In such an environment, breathing air containing only 0.1% carbon monoxide causes 50% of the hemoglobin in the blood to convert into carboxyhemoglobin within an hour. Half an hour more exposure to this environment makes death inevitable (18).

Symptomatology in poisoning can occur monophasically or biphasically. In the monophasic condition, patients can recover from poisoning with or without sequelae, with functional and tissue impairment. In the biphasic form, a period of coma follows, making the patient and treatment difficult. There is an interval period. This is a pseudo-period in which the patient's clinical condition remains good, and his consciousness remains temporarily clear. After this pseudoperiod, which can mislead the physician, the patient's clinical condition deteriorates rapidly, and the patient dies (19).

Since the poisoning mechanism is based on blood proteins such as erythrocytes and every organ requires blood supply, all organs in the clinic are affected by this poisoning. However, the main symptoms are cardiovascular and neuropsychiatric since the brain and heart are organs with high oxygen consumption (20). The brain regulates its respiration according to the carbon monoxide concentration in the blood. Therefore, people who are unaware they are breathing carbon monoxide will not notice anything wrong. At the same time, he will continue to breathe normally until he loses consciousness. In case of death, the skin becomes pinkish at autopsy, and the blood is bright red due to carboxyhemoglobin (21).

Although the typical sign of poisoning is the appearance of cherry red lips, this is a non-specific sign. According to the observations made by the Düzce Faculty of Medicine, patients who experienced acute poisoning:

**Respiratory system:** Exercise dyspnea, signs of upper respiratory tract infections

**Nervous system:** Lethargy, confusion, depression, hallucination, agitation, syncope, coma, memory and gait disturbance, dizziness, headache

Clinical findings such as chest pain, sinus tachycardia, and palpitations have been observed. It should not be overlooked that the symptoms are mainly neuropsychiatric and cardiovascular and are not specific(22).

However, since carboxyhemoglobin levels may be at values that may mislead the physician, the disease history should support the results. What is important here is the blood exposure time, exposure amount, respiratory rate, and the amount of oxygen in the environment. Carboxyhemoglobin levels may decrease further in the long term. Low carboxyhemoglobin levels can also lead to poisoning and put the patient's life in danger (9).

## Diagnosis

Poisoning is often confused with viral upper respiratory tract disease, and when acute symptoms are considered, it is seen that the patient does not have any specific findings. For this reason, the most crucial point in diagnosis is the history of poisoning (5).

During the physical examination, the patient had cherry red lips, pale skin, and retinal Flame burn in the eye due to hemorrhage, the retinal bright red color of the veins (a sensitive early finding), papillary edema, loss of vision in both eyes simultaneously, pupillary Findings such as mydriatic may be observed. However, these are also non-specific findings (23,24).

In cases of poisoning, oxygen saturation can be measured with a pulse oximeter. However, since the measurement deals with the respiratory system, it may be expected in patients whose respiratory system is unaffected. It may give misleading results to the physician. For this reason, using a pulse oximeter in diagnosis may cause possible poisoning to be overlooked. Also, pulse oximeter does not provide information about carboxyhemoglobin and oxyhemoglobin levels. The primary mechanism of poisoning is oxygen transport in hemoglobin. Since carbon monoxide is the inhibitor, correct diagnosis requires measuring carboxyhemoglobin levels (25,26).

*Carboxyhemoglobin in poisoning is diagnosed from a heparin-treated blood sample (arterial or venous); arterial blood is preferred (27). Symptoms, which even the individual can hardly notice, begin at a concentration of 3%, and the findings become more severe when they reach 10%. It is realized that physiological abnormalities might occur. However, applying this test to smokers after smoking may give misleading results. In cigarette addiction, the carboxyhemoglobin value can reach approximately 5.5% for 20 cigarettes (1 pack) per day. (5,28). While the carboxyhemoglobin level is around 1% in non-smoking individuals, this value is up to 15% in patients with a history of heavy smoking (29). In general, carboxyhemoglobin levels below 40% are not associated with coma or death. A carboxyhemoglobin concentration of 40% and above is observed in high toxications. In 20% of cases, breathing becomes more complex, and the patient suffers headaches and dizziness. A carboxyhemoglobin level of 5% is the concentration at which oxygen transport is inadequate; hemolytic anemia might occur. The first signals that treatment should be started should be received here (30).*

Considering all those symptoms and values, the patient has comorbid serum lactate increment and metabolic abnormalities. If there is acidosis, it should be regarded as that the patient may be exposed to long-term carbon monoxide (25).

Hemogram tests for immunity and tissue hypoxia, liver function tests for carbon monoxide metabolism, muscle and cardiac enzyme tests to see the effect it creates by binding to myoglobin, and urinalysis, a measure of blood, should be requested from the laboratory. In case of clinical suspicion, chest radiography, computerized brain tomography, magnetic resonance and electrocardiogram methods should be used. However, after these tests are conducted, the time it takes for the results to outcome should be taken into consideration. The patient should take necessary symptomatic treatment (31).

## Treatment

Treatment can proceed symptomatically. If hypotension occurs, the patient and the serum isotonic are placed in the Trendelenburg position. Infusion is made, and vasoconstrictor drugs can be used if necessary. If there is persistent hypotension, noradrenaline treatment should be started in the patient. Patients with neurogenic seizures should be treated with benzodiazepine. If there is no response or the seizure recurs, other antiepileptics, such as barbiturates, are added to the treatment. Barbiturates should be used with caution as the somnolence threshold level will be lower than benzodiazepines. If fever develops due to activation of the immune system, intravenous paracetamol should be started as an antipyretic. Possible use of NSAIDs will cause constriction in the afferent artery in the kidney, which will reduce blood flow in the kidney, thus causing acute renal failure (5).

Protocols that significantly increase patient survival. The first is hyperbaric oxygen therapy, and the other is oxygen intake through a mask.

**Oxygen therapy with mask:** 4-6 hours until the patient's clinical findings improve or until the COHb level drops below 5% in mild/moderate poisoning. In severe poisoning, it should be given until the COHb level drops below 2% (32).

**Hyperbaric oxygen therapy:** *Separation of carbon monoxide bound to hemoglobin is impossible with pure oxygen gas; therefore, high-pressure oxygen must be used in treatment. Hyperbaric oxygen therapy can quickly improve symptoms and reduce deaths due to poisoning when administered within the first 6 hours. In addition, it is a more effective method than normobaric oxygen therapy in preventing neuropsychiatric symptoms that occur with chronic toxic effects as well as in the acute period (33). Hyperbaric oxygen therapy should be preferred in patients experiencing loss of consciousness (34). Hyperbaric oxygen therapy has shown its success even in neuropsychiatric symptoms that appeared after three weeks (35).*

## CONCLUSION

The first symptoms of carbon monoxide poisoning are nonspecific. Physical examination findings are limited in diagnosis. The patient's epicrisis is the most trusted diagnostic tool. When poisoning is suspected, the COHb level in the blood should be measured as soon as possible. The source of carbon monoxide in the patient's environment taken by ambulance should be identified, and the source should be eliminated. The patient should be provided with oxygen support, and hospitalization should continue. Providing oxygen support to a patient brought to the emergency room by ambulance without the knowledge of the physician causes low carbon monoxide levels in the blood. In this case, the physician must obtain the necessary information from the healthcare professional who provides first aid. Otherwise, disruption of the treatment is inevitable (36,37). There is no chemical or physiological antidote for poisoning. The mechanism of toxicity should be aimed at separating carbon monoxide from protein (38).

## DESCRIPTIONS

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